

Heat and Mass Transfer (DI05019071)

Complete Lecture Deck

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July 23, 2026

Effectiveness of Heat Exchanger: Applications

- Course: Heat and Mass Transfer (DI05019071)
- Unit 4: Heat Exchanger
- Lecture 36: Applying the NTU Method

Lecture Agenda

- 1. Reading Effectiveness-NTU Charts
- 2. Special Case: Phase Change ($c = 0$)
- 3. Special Case: Balanced Flow ($c = 1$)
- 4. Step-by-Step: Performance Problems
- 5. Step-by-Step: Design/Sizing Problems
- 6. Unit 4 Summary

Effectiveness-NTU Charts

- Instead of complex formulas, engineers often use charts.
- X-axis: NTU (Number of Transfer Units)
- Y-axis: Effectiveness (ϵ , from 0 to 1 or 0 to 100%)
- Curves: Different values of capacity ratio 'c' (from 0 to 1).

Chart Observations

- For a given NTU and given flow type, effectiveness increases as ' c ' decreases.
- Maximum effectiveness always occurs at $c = 0$.
- For NTU ≥ 3 , the curves flatten out.
- Increasing NTU (making the exchanger larger/more expensive) yields diminishing returns in performance.

Special Case: Phase Change ($c = 0$)

- For condensers and evaporators, $C_{\max} = \text{infinity}$, so $c = 0$.
- Looking at the formulas or charts, the flow arrangement (parallel vs counter) disappears from the math.
- $\varepsilon = 1 - \exp(-NTU)$ (for all flow arrangements!)
- This perfectly matches our conclusion from the LMTD method.

Special Case: Balanced Flow ($c = 1$)

- If $C_{\min} = C_{\max}$, then $c = 1$.
- Example: Gas-to-gas heat recovery with equal flow rates.
- Parallel Flow: $\varepsilon = 0.5 * [1 - \exp(-2*NTU)]$ (Max $\varepsilon = 50\%$)
- Counter Flow: $\varepsilon = NTU / (1 + NTU)$ (Max ε approaches 100%)
- This proves counter flow is vastly superior for balanced flows.

Solving Performance Problems

- Goal: Find outlet temps when Area (A) is given.
- 1. Calculate C_h , C_c . Determine $C_{\min}$, $C_{\max}$, and c .
- 2. Calculate $NTU = UA / C_{\min}$.
- 3. Find ε using formulas or charts.
- 4. Calculate $q_{\max} = C_{\min} * (T_{h,in} - T_{c,in})$.
- 5. Calculate actual $q = \varepsilon * q_{\max}$.
- 6. Calculate $T_{h,out}$ and $T_{c,out}$ using q .

Solving Design (Sizing) Problems

- Goal: Find Area (A) when inlet and outlet temps are known.
- 1. Calculate C_h , C_c . Determine C_{min} , C_{max} , and c .
- 2. Calculate actual q , and q_{max} .
- 3. Calculate required $\varepsilon = q / q_{max}$.
- 4. Find required NTU using charts (or inverted formulas).
- 5. Calculate Area = $(NTU * C_{min}) / U$.

Example Problem: Performance

- A counter-flow exchanger has $U = 200 \text{ W/m}^2\text{K}$, $A = 10 \text{ m}^2$.
- Hot water: 2 kg/s , $c_p = 4180 \text{ J/kgK}$, inlet = 90°C .
- Cold water: 1 kg/s , $c_p = 4180 \text{ J/kgK}$, inlet = 20°C .
- Step 1: $C_h = 8360 \text{ W/K}$. $C_c = 4180 \text{ W/K}$. $C_{\min} = C_c = 4180$.
- Step 2: $c = 4180/8360 = 0.5$.

Example Problem: Solution

- Step 3: $NTU = UA/C_{\min} = (200 \cdot 10)/4180 = 0.478$.
- Step 4: Use chart or formula for Counter flow ($NTU=0.478$, $c=0.5$) - $\epsilon \approx 0.35$
- Step 5: $q_{\max} = 4180 \cdot (90 - 20) = 292,600 \text{ W}$
- Step 6: $q_{\text{actual}} = 0.35 \cdot 292,600 = 102,410 \text{ W}$
- Step 7: $T_{c,\text{out}} = 20 + 102,410/4180 = 44.5^\circ\text{C}$

Choosing Between Methods

- When to use LMTD:
 - - Sizing a new heat exchanger (finding Area).
 - - All inlet and outlet temperatures are known or easily found.
- When to use Effectiveness-NTU:
 - - Evaluating an existing heat exchanger.
 - - Finding outlet temperatures.
 - - Comparing different heat exchanger designs (ϵ is a great metric).

Unit 4 Summary

- Unit 4 covered the analysis of heat exchangers.
- We learned basic classifications (parallel, counter, cross flow).
- We defined the Overall Heat Transfer Coefficient (U) and fouling.
- We derived and applied the LMTD method.
- We defined Effectiveness, NTU, and applied the NTU method.

Next Unit Preview

- Unit 5: Mass Transfer
- We will move away from moving heat, and look at moving mass.
- Topics will include Fick's Law of Diffusion, concentration gradients, and modes of mass transfer.
- Prepare by reviewing concentration definitions (molarity, mass fraction).