

Heat and Mass Transfer (DI05019071)

Complete Lecture Deck

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July 23, 2026

Effectiveness of Heat Exchanger and NTU Method

- Course: Heat and Mass Transfer (DI05019071)
- Unit 4: Heat Exchanger
- Lecture 35: Effectiveness-NTU Method (Part 1)

Lecture Agenda

- 1. Why not use LMTD?
- 2. Heat Capacity Rate (C)
- 3. Maximum Possible Heat Transfer ($q_{\max}$)
- 4. Heat Exchanger Effectiveness (ε)
- 5. The Number of Transfer Units (NTU)
- 6. Capacity Ratio (c)

The Limitation of LMTD

- LMTD is excellent for 'Sizing' problems:
 - - You know inlet and outlet temps, you find LMTD, then find Area.
- LMTD is terrible for 'Performance' problems:
 - - You have a specific heat exchanger (Area known).
 - - You know inlet temps, but NOT outlet temps.
 - - Finding outlet temps requires guessing them, calculating LMTD, and iterating until things match.

Heat Capacity Rate (C)

- First, we define the Heat Capacity Rate, C.
- $C = \text{mass flow rate (m)} * \text{specific heat (cp)}$
- $C_h = m_h * cp_h$ (Hot fluid capacity rate)
- $C_c = m_c * cp_c$ (Cold fluid capacity rate)
- Units: W/K or J/s·K

Identifying $C_{\min}$ and $C_{\max}$

- Calculate both C_h and C_c .
- The smaller value is designated as $C_{\min}$.
- The larger value is designated as $C_{\max}$.
- The fluid with $C_{\min}$ will experience the LARGER temperature change (ΔT) in the heat exchanger.
- Since $q = C * \Delta T$, if C is small, ΔT must be large for the same q .

Maximum Possible Heat Transfer ($q_{\max}$)

- What is the theoretical maximum heat transfer possible?
- It happens in an infinitely long counter-flow heat exchanger.
- The fluid with $C_{\min}$ will reach the inlet temperature of the other fluid.
- $q_{\max} = C_{\min} * (T_{h,in} - T_{c,in})$
- The max temperature difference is $(T_{h,in} - T_{c,in})$.

Defining Effectiveness (ε)

- Effectiveness (ε) is the ratio of actual heat transfer to maximum possible heat transfer.
- $\varepsilon = q_{\text{actual}} / q_{\text{max}}$
- Effectiveness is dimensionless and $0 \leq \varepsilon \leq 1$.
- $q_{\text{actual}} = \varepsilon * C_{\text{min}} * (T_{\text{h,in}} - T_{\text{c,in}})$

Calculating Actual Heat Transfer

- Once q_{actual} is found via effectiveness, outlet temps are easy:
- $q_{\text{actual}} = C_h * (T_{h,\text{in}} - T_{h,\text{out}})$
- $T_{h,\text{out}} = T_{h,\text{in}} - (q_{\text{actual}} / C_h)$
- $q_{\text{actual}} = C_c * (T_{c,\text{out}} - T_{c,\text{in}})$
- $T_{c,\text{out}} = T_{c,\text{in}} + (q_{\text{actual}} / C_c)$

How to find Effectiveness?

- Effectiveness depends on the geometry of the heat exchanger and two dimensionless groups.
- $\varepsilon = \text{function}(\text{NTU}, c, \text{flow arrangement})$
- We need to define NTU and c to use the formulas and charts.

Number of Transfer Units (NTU)

- NTU is a dimensionless parameter indicating the 'size' or heat transfer capability.
- $NTU = (U * A) / C_{min}$
- Higher NTU means a larger heat exchanger (more Area or better U).
- Practically, for $NTU \geq 3$, the size of the heat exchanger grows massively for very small gains in effectiveness.

Capacity Ratio (c)

- Capacity ratio ' c ' is the ratio of the minimum to maximum heat capacity rates.
- $c = C_{\min} / C_{\max}$
- Since $C_{\min} \leq C_{\max}$, the ratio c must be: $0 \leq c \leq 1$.
- If $c = 0$, one fluid is undergoing phase change ($C_{\max} = \text{infinity}$).
- If $c = 1$, both fluids have the same capacity rate (balanced flow).

Effectiveness Formulas Preview

- Parallel Flow: $\varepsilon = [1 - \exp(-NTU*(1+c))] / (1+c)$
- Counter Flow ($c \neq 1$): $\varepsilon = [1 - \exp(-NTU*(1-c))] / [1 - c*\exp(-NTU*(1-c))]$
- Counter Flow ($c=1$): $\varepsilon = NTU / (1 + NTU)$
- These are standard formulas provided in data books.

Summary & Next Lecture

- LMTD requires iteration for performance problems.
- $q_{\max}$ occurs when the $C_{\min}$ fluid reaches the opposite inlet temperature.
- Effectiveness $\varepsilon = q_{\text{actual}} / q_{\max}$.
- ε depends on NTU ($UA/C_{\min}$) and c ($C_{\min}/C_{\max}$).
- Next Lecture: Using Effectiveness charts and solving numerical problems.